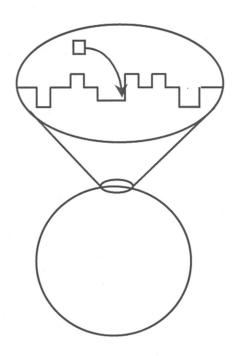
Nucleation Energy Barriers and the Coarsening of Faceted Crystals

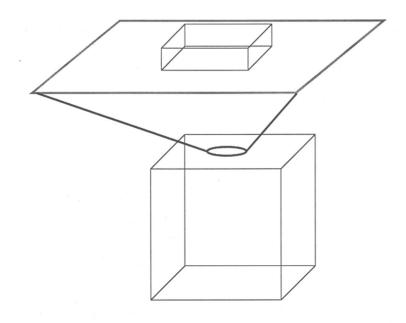
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Growth Resistance



On a "rough surface," there is no barrier to the attachment of atoms (assumption in LSW theory).

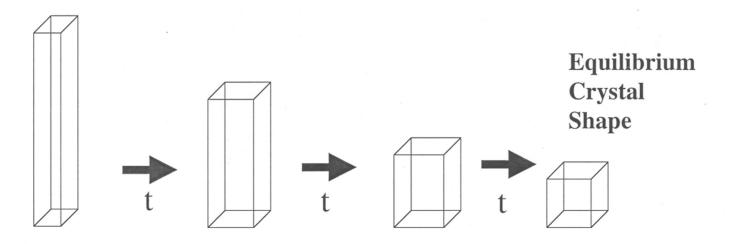


On a facet, there is a nucleation barrier. The creation of new edges must be balanced by the volume energy gained by reducing the chemical potential of the material that condenses.

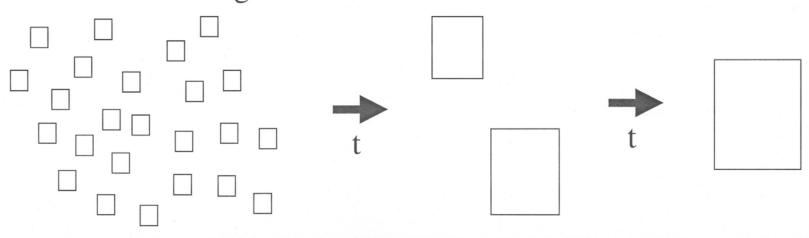
Burton, Cabrera, Frank, Philos. Trans. R. Soc. London, Ser. A, 243 300-58 (1951).

Situations with Low Supersaturation

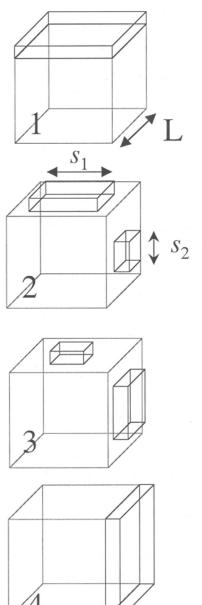
1. Morphological Evolution



2. Coarsening

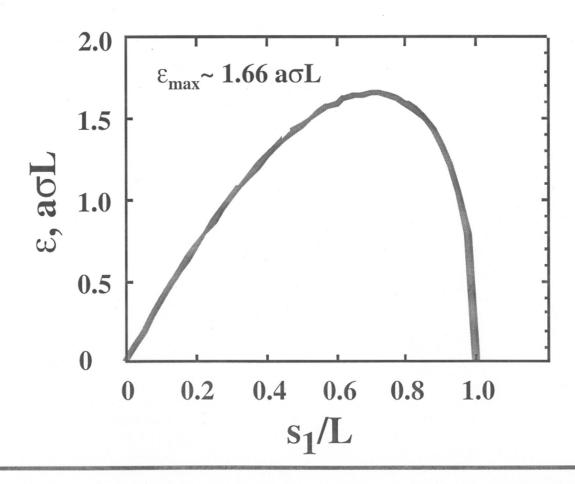


The Barrier to Shape Changes: Cube



$$\varepsilon(s_1) = 4a\sigma s_1 + 4a\sigma s_2 - 4a\sigma L$$

$$\varepsilon(\mathbf{s}_1) = 4a\sigma s_1 + 4a\sigma (L^2 - s_1^2)^{1/2} - 4a\sigma \mathbf{L}$$



Size of the Nucleation Energy Barrier

$$\epsilon_{max}$$
~ 1.66 a σ L

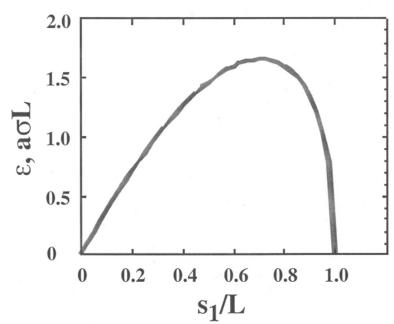
$$Let S = 1 \mu m$$

$$a = 2.5 \text{ Å}$$

$$\sigma = 1 \text{ J/m}^2$$

$$kT = 10^{-20} \text{ J}$$

$$\varepsilon_{\text{max}}$$
= 1.66a σ S = 4 x 10⁴ kT



• Barriers larger than 60kT are never surmounted

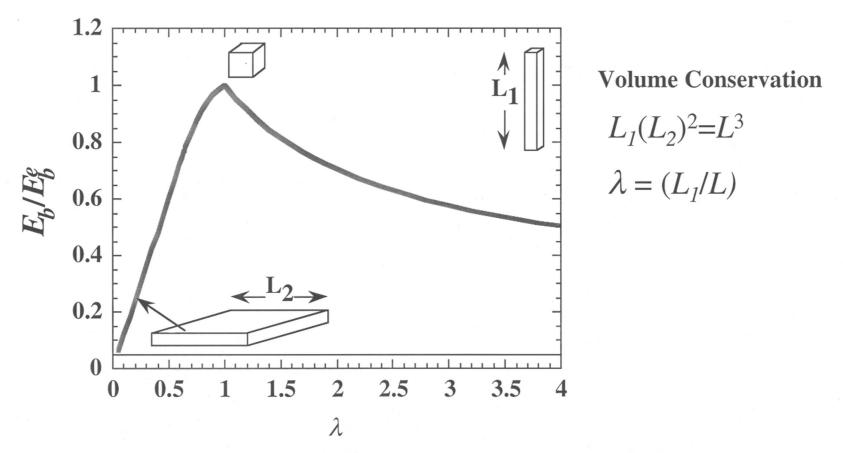
Let
$$S = 1 \text{ nm}$$

 $a = 2.5 \text{ Å}$
 $\sigma = 1 \text{ J/m}^2$
 $kT = 10^{-20} \text{ J}$

$$\epsilon_{max}$$
= 1.66a σ S = 40 kT

• Only "nanocrystals" are able to provide equivalent supersaturations necessary for island nucleation.

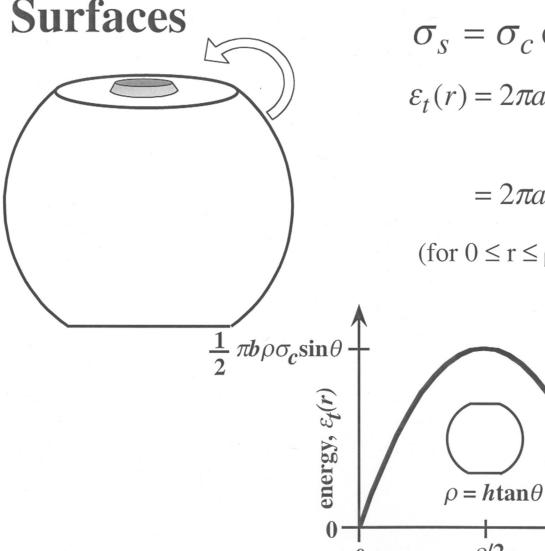
Nonequilibrium Shapes: Cube



The nucleation energy barrier remains significant even for large departures from the equilibrium crystal shape.

The Barrier on Shapes with Rough



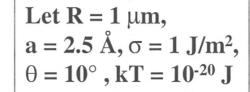


$$\sigma_s = \sigma_c \cos \theta$$

$$\varepsilon_t(r) = 2\pi a r \sigma_c \sin \theta - \pi a r^2 \left(\frac{2\sigma_c}{R}\right)$$

$$=2\pi a\sigma_c\sin\theta\left(r-\frac{r^2}{\rho}\right)$$

(for
$$0 \le r \le \rho$$
)



$$E_b = 1 \times 10^3 \text{ kT}$$

Let
$$R = 10 \text{ nm}$$

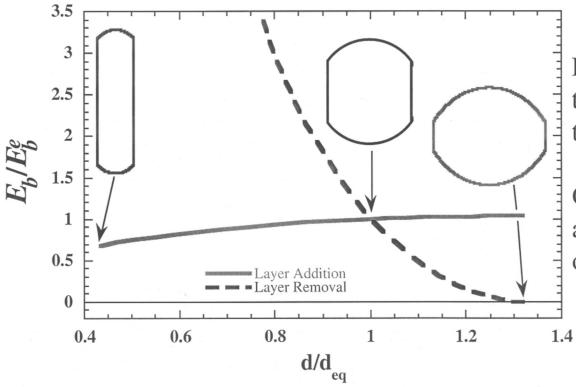
 $E_b = 10 \text{ kT}$

radius of nucleus, r

 $\rho/2$

Nonequilibrium Truncated Sphere

$$\begin{split} E_b(+)/E_b^e &= 2/R\kappa \\ E_b(-)/E_b^e &= (2/R\kappa)[R\kappa(\rho/\rho_e)-1]^2 \end{split}$$

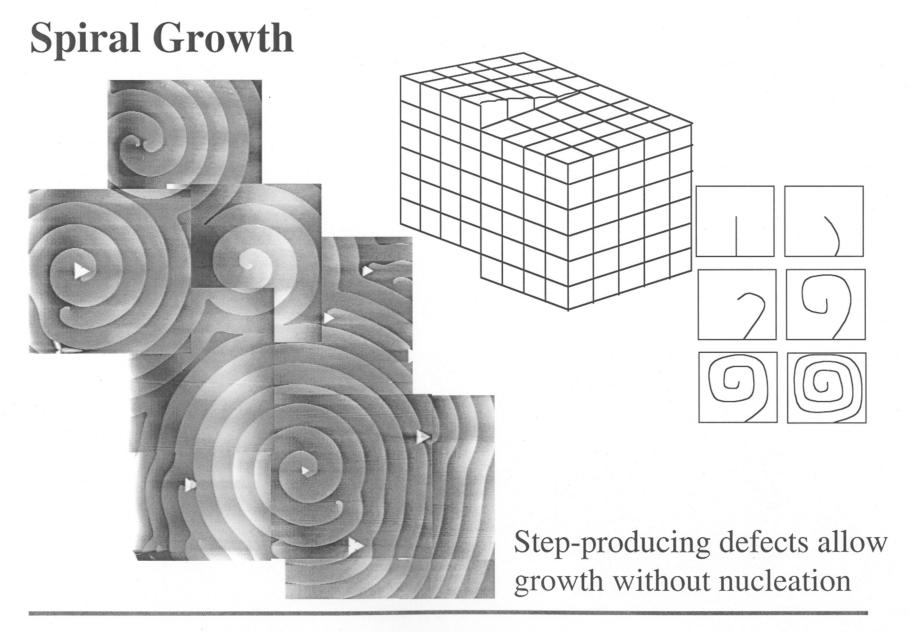


Prolate: facet moves towards the center of the crystal.

Oblate: facet moves away from the center of the crystal.

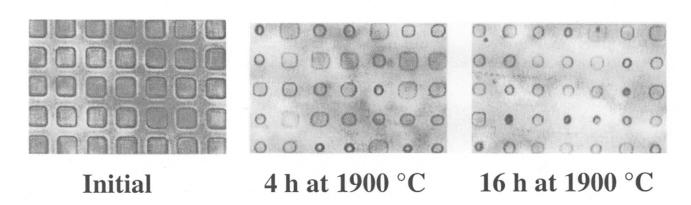
- Assume constant volume
- Assume constant curvature
- Assume that the contact angle is preserved

How do Real Crystals Grow?



Evidence for the Nucleation Energy Barrier and Defect Controlled Evolution

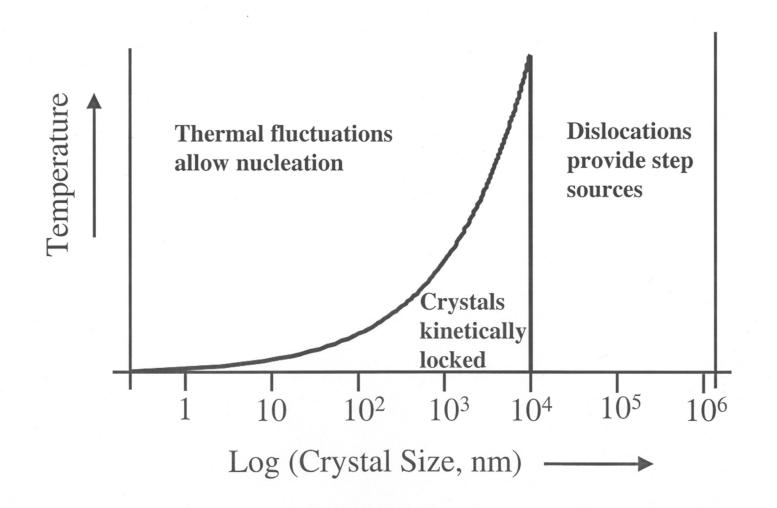
20 μm pores in Al_2O_3 oblate shapes with large (10 $\overline{1}$ 2) facets



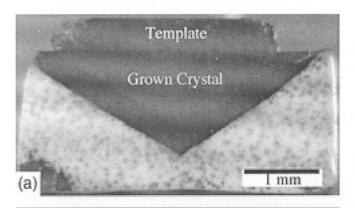
Kitayama, Narushima, Glaeser, *J. Am. Ceram. Soc.*, 83 [10] 2572-82 (2000).

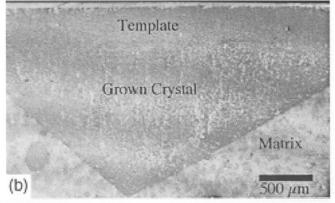
Some pores evolve to the ECS, others are stuck.

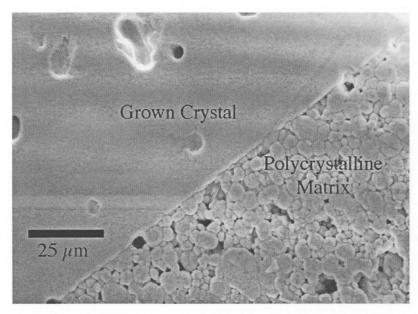
Shape Change Kinetics Depend on Size and Defect Structure (Schematic)



Size Effects are also Observed in Coarsening



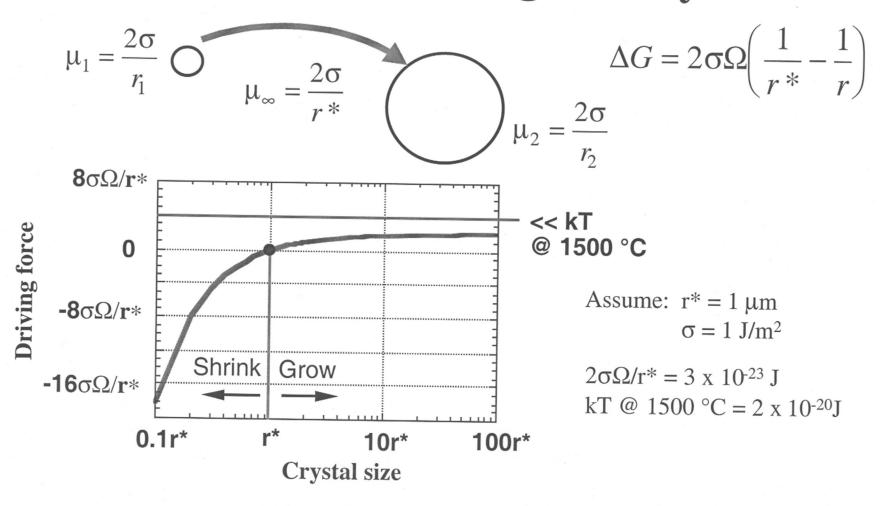




P.W. Rehrig, G.L. Messing, S. Trolier-McKinstry, "Templated Grain Growth of Barium Titanate Single Crystals,"
J. Am. Ceram. Soc., 83 [11] 2654–60 (2000)

- Growth occurs in a eutectic liquid
- Seed growth 790 μ m/h, matrix grains \leq 9 μ m/h

Conventional Coarsening Theory



- LSW assumes growth/dissolution occurs by transport along capillarity-induced chemical potential gradients.
- Driving forces for growth are small compared to kT

The Nucleation Energy Barrier

Energy barrier for crystal at r*

$$\varepsilon(s) = 4as\sigma - as^2 \frac{2\sigma}{r^*}$$
$$\varepsilon_+ = 2a\sigma r^*$$

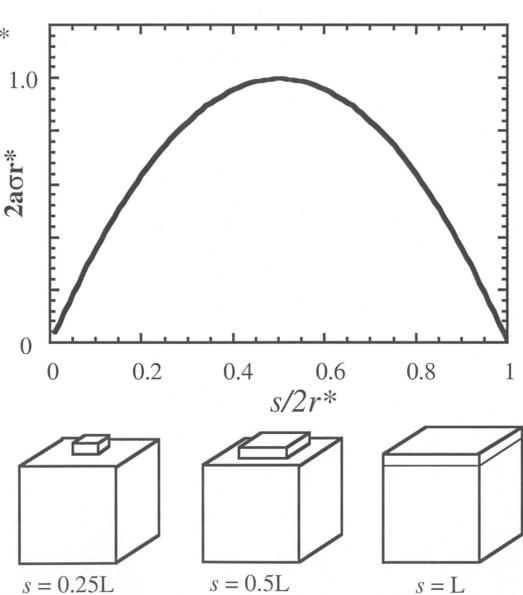
$$\varepsilon_{+} = 2a\sigma r *$$

$$(L = 2 r^*)$$

$$a = 2.5 \times 10^{-10} \text{m}$$

 $r^* = 1 \times 10^{-6} \text{m}$
 $\sigma = 1 \text{ J/m}^2$
 $\epsilon_+ = 5 \times 10^{-16} \text{ J} >> \text{kT}$

Barriers greater than 40kT are not surmounted



NEB as a Function of Crystal Size

Layer removal

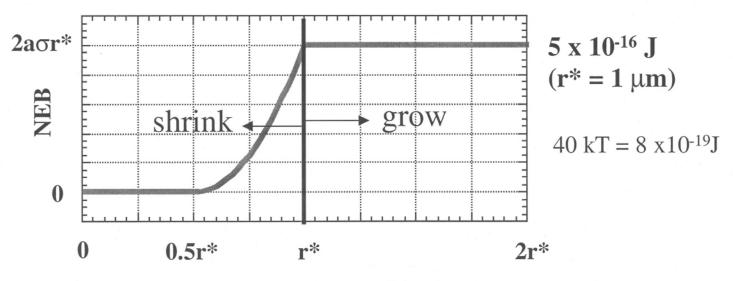
$$\varepsilon(s) = 4as\sigma - 4aL\sigma + (aL^2 - as^2)\frac{2\sigma}{r^*}$$

$$\varepsilon_{-} = 2a\sigma \left[R^* + L \left(\frac{L}{r^*} - 2 \right) \right]$$

Layer addition

$$\varepsilon(s) = 4as\sigma - as^2 \frac{2\sigma}{r^*}$$

$$\varepsilon_{+} = 2a\sigma r *$$



Crystal Size (L)

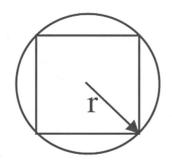
Thermal fluctuations can not support 2D nucleation during coarsening of micron-sized crystals

Model for Coarsening of Faceted Crystals

For crystals of all sizes nucleation rate = rate material arrives at surface by diffusion

Crystal is in equilibrium with a chemical potential of

$$\mu_e = \frac{2\sigma}{r}$$



Potential selected to balance diffusion and nucleation rates

$$\mu_e < \mu_s < \mu_\infty$$

Mean field potential

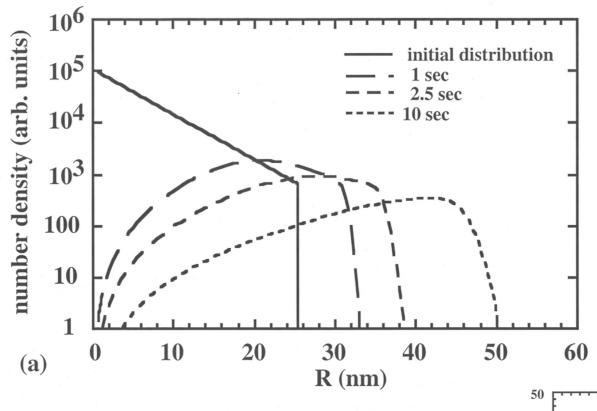
$$\mu_{\infty} = \frac{2\sigma}{r^*}$$

For a starting distribution, n(R), numerically determine r^* and local chemical potentials (μ_s) for each crystal size class, under the constraint that the total volume is conserved.

Examine three cases:

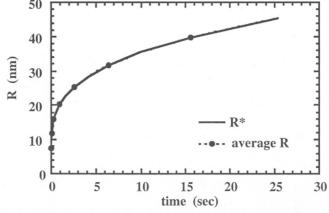
- 1. No crystals have a NEB; LSW
- 2. All crystals have a NEB
- 3. Most crystals have a NEB, others do not.

No Barrier to Nucleation

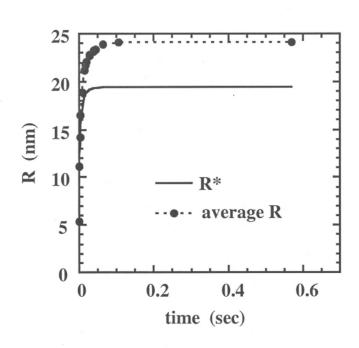


The barrier is "turned off" by lowering the surface energy to 0.001 J/m².

• Results are consistent with LSW theory

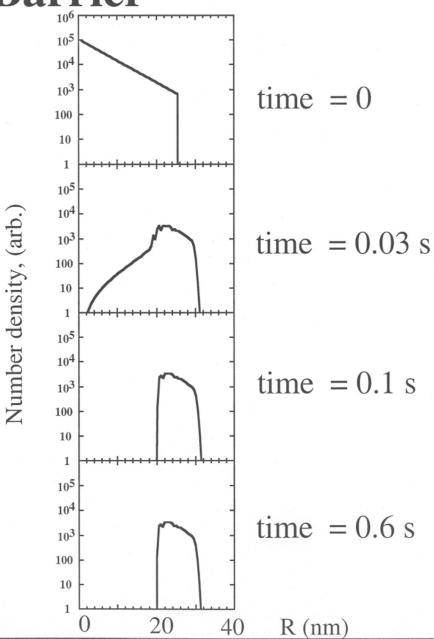


Coarsening with a Barrier

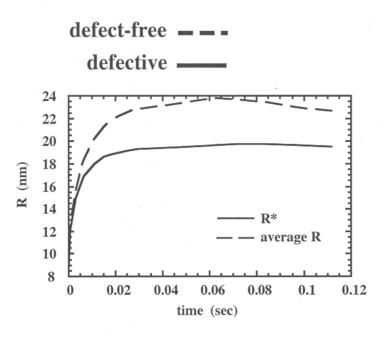


Distribution becomes frozen above a critical size where thermal fluctuations are insufficient to drive nucleation.

$$\sigma = 0.1 \text{ J/m}^2$$

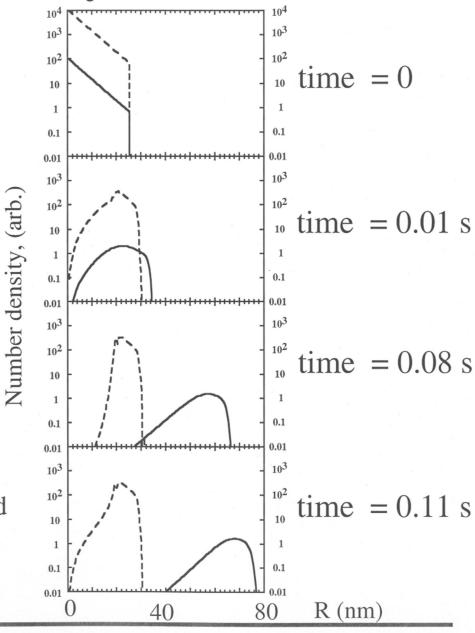


Ideal and Defective Crystals



• Crystals with a persistent step source grow at the expense of ideal crystals

• A bimodal distribution is developed



Predictions for Faceted Crystal Coarsening

- In populations containing a mixture of prefect and defective crystals, the crystals with step sources grow at the expense of the perfect ones.
- When the ideal crystals have been consumed, the remaining crystals no longer have an advantage, and coarsening should proceed normally.
- Large seeds, added during TGG and SSCC processing, have a higher probability of containing a defect and grow at the expense of the matrix.

Conclusions

- In systems with micron-scale crystals, capillary driving forces and thermal fluctuations are insufficient to sustain nucleation on flat surfaces; the coarsening of faceted crystals beyond this size must be defect mediated
- In populations of real crystals containing a mixture of perfect and defective particles, mass will accumulate on defective particles.
- If step producing defects are relatively rare, then abnormal coarsening will occur. If step producing defects are relatively common, then normal coarsening will occur.